

Intermediate Chemistry 13-15 Age Category: 13-15

- 1. Bonding, structure, and the properties of matter
- 2. Electromagnetic radiation, photoelectric effect,
- 3. The Bohr model, (Z = 1) or for a hydrogen-like ion (Z > 1)
- 4. The quantum mechanical model of an atom, quantum numbers, orbital shapes and energies
- 5. Units of measurement, significant figures, uncertainty in measurement, dimensional analysis
- 6. Temperature, Density, Pressure, and Classification of matter
- 7. Organic chemistry Introduction to Advanced
- 8. Fundamental laws of chemistry (law of conservation of mass, law of definite proportion, law of multiple proportion), Dalton's atomic law
- 9. Electromagnetic radiation, photoelectric effect, the Bohr model, the quantum mechanical model of an atom, quantum numbers, orbital shapes and energies
- 10. The rate and extent of chemical change
- 11. The Periodic Table, Aufbau principle, periodic trends in atoms
- 12. Acids and bases, chemical reactions and equations, naming simple inorganic compounds
- 13. Quantitative chemistry
- 14. Atomic mass, Moles, Molecules and ions, molar mass
- 15. Types of chemical bonds, bond polarity, covalent chemical bond
- 16. Solutions and mixtures, calculating solution composition: molarity, molarity, and weight percent

Industry Trips to see Chemistry and Chemists at work

Trip to Francis Crick Institute
Trip to Science museum
Afternoon trip to zoology museum
Trip to London Royal Hospital
Trip to Cancer Institute
TBC 2019 probably Trip to Chemical GlaxoSmithKline - Brentford HQ

Trip to Colleges in London and Cambridge University

Kings College Cambridge, Corpus Christie Cambridge , Sydney Sussex Cambridge UCL London, Imperial London, LSE London, Kings College London